

Silver in Wound and Trophic Ulcer Treatment: A Modern View of the Problem

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Abstract

This review presents published data on the history of use, mechanism of action, and effectiveness of silver and silver-based drugs in surgical practice. A literature search was carried out using PubMed, EMBASE, Google Scholar, and E-Library databases. Analysis of available literature data convincingly demonstrates the effectiveness of silver nanocomposites as antibacterial and anti-inflammatory agents. The demonstrated antimicrobial property of silver nanoparticles (AgNPs) against various antibiotic-resistant bacteria is especially significant for clinical use. Silver nanoparticles have clinically proven effective in treating wounds and trophic ulcers. (**International Journal of Biomedicine. 2024;14(2):240-245.**)

Keywords: silver • silver nanoparticles • wound • trophic ulcer

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Abbreviations

Ag, silver; **AgNPs**, silver nanoparticles; **DNA**, deoxyribonucleic acid; **GSH**, reduced glutathione; **ROS**, reactive oxygen species.

Introduction

Silver (Ag) is one of the metals most intensively used by humanity for medical purposes since ancient civilizations. The use of silver in medicine is based primarily on its disinfectant properties since silver has the most potent bactericidal effect among metals.⁽¹⁾ The bactericidal effect of Ag-based compounds is 1750 times more powerful than the effect of the same concentration of carbolic acid, 1500 times higher than the effect of the same concentration of phenol, and 3.5 times stronger than the effect of sublimate. Silver, having a powerful antimicrobial effect, is detrimental to antibiotic-resistant strains of more than 500 species of bacteria.⁽²⁾ It has been established that Ag has the most powerful bactericidal effect among heavy metals.⁽³⁾

Hippocrates, the father of modern medicine, believed that silver powder has healing properties and is recommended

for treating trophic ulcers.^(1,4) One of the first silver compounds used in medical practice was silver nitrate. It was originally used in a solid form known as lapis, or hellstone.⁽¹⁾ In the 19th century, lapis became widespread for cauterizing excessive granulations in wounds and treating burns, and Ag-based compounds became one of the main means for the treatment and prevention of wound infections before the invention of antibiotics.⁽⁵⁾ It was in the 19th century that active scientific study of silver and its compounds began, which led to the discovery of its antiviral, antibacterial, and immunomodulatory activities.⁽⁶⁾ From the end of the 19th to the beginning of the 20th century, several Ag-based compounds and drugs were developed: collargol, protargol, elargol, silargel, argosulfan, and others. Some of them are still successfully used today.⁽¹⁾

After the discovery of penicillin, the first antibiotic, in 1941, interest in silver as an antimicrobial drug disappeared for almost 40 years. Some experts believe that before the discovery of antibiotics, silver salts were one of the most widely used agents with antimicrobial activity.⁽⁷⁾ However, the emergence of antibiotic-resistant strains of microorganisms has led to the search for new silver-based antibacterial drugs. In 1968, silver

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sulfadiazine was introduced into clinical practice and began one of the most successful silver-containing antimicrobials. Since that time, topical silver sulfadiazine ointment has been the standard antibacterial treatment for major burns and is widely used in their therapy.⁽⁸⁾ The therapeutic effect of silver sulfadiazine in the local treatment of wounds and wound infections is associated with its anti-inflammatory and antiseptic effects, as well as stimulation of the proliferation and differentiation of keratinocytes.⁽⁹⁾

Currently, one of the most effective topical Ag-based drugs for the local treatment of wounds and wound infections is silver sulfathiazole. According to immunohistochemical analysis, when using silver sulfathiazole for local therapy in the complex treatment of complicated diabetic foot syndrome, there is a decreased expression of CD68⁺ macrophages in the studied tissues of trophic ulcers and wounds. In a study by Kalinichenko et al.,⁽¹⁰⁾ a gradual increase in the coefficient of macrophage activity in biopsy specimens, with the systematic achievement of a threshold level of 1.0, was a reliable sign of a better regenerative response and indicated the optimal nature of the course of the wound process. The use of silver sulfathiazole in the complex treatment of patients with diabetic foot syndrome reduced the intensity of inflammatory reactions and the content of C-reactive protein and neopterin,⁽¹¹⁾ and it had a protective effect on the vascular endothelium.⁽¹²⁾

In recent years, various silver-based dressings have been proposed for treating wounds (Silvercel, Silverlon, Silvasorb, Contreet-H, Arglaes, Aquacel-Ag, Acticoat, Nucryst, and others). Such dressings have a more practical application than metallic silver or silver salts.⁽¹³⁾ Several studies have noted a significant reduction in the time of wound cleansing and healing, in elimination of systemic manifestations of the infectious process, in duration of required antibacterial therapy, and in hospitalization in patients using silver-based dressings. Using silver-based dressings in a moist microenvironment creates optimal conditions for epithelization during the wound process.⁽¹⁴⁾

It is now generally accepted that silver is a broad-spectrum antimicrobial agent. Because of the broad spectrum of antibacterial action, from bacteriostatic to bactericidal, Ag-based compounds have found wide applications in medicine.^(15,16) In 2009, approximately 15 tons of silver were included in medical products worldwide.⁽¹⁷⁾

New prospects for the medical use of silver are opening in connection with the production of its most unique form—silver nanoparticles (AgNPs), which have more outstanding physicochemical and biological properties beyond bulk silver.⁽¹⁸⁾ AgNPs can destroy multiple drug-resistant strains and prevent biofilm formation, indicating significant potential in antibacterial application.⁽¹⁹⁾

The antibacterial mechanisms of AgNPs have been discussed extensively, but the exact effect of AgNPs on bacteria is still undefined.⁽²⁰⁾ However, two mechanisms attract attention, namely contact killing and ion-mediated killing.

Ionized silver can “attack” at least three cellular components: cell membranes, cytoplasmic organelles, and DNA.⁽²¹⁾ Silver ions interact with membrane proteins and further block the respiratory chain.⁽²²⁾ It has been reported that AgNPs can anchor to the bacterial cell wall, causing physical

changes in the bacterial membrane followed by its damage, which can lead to leakage of cellular contents and bacterial death.⁽²³⁻²⁵⁾ After adhesion to the bacterial wall, AgNPs can also penetrate the bacteria through the membrane. Small nanoparticles, which have a larger surface area in contact with bacterial cells, can reach the cytoplasm more often than larger nanoparticles.⁽²⁴⁾ Once inside the microbial cell, AgNPs can interact with cellular structures and biomolecules such as proteins, lipids, and DNA, thereby leading to bacterial dysfunction and, ultimately, death.⁽²⁶⁻²⁹⁾ It is also speculated that AgNPs interact effectively with the carboxyl and thiol groups of β -galactosidase and inhibit intracellular biological functions, leading to cell death.⁽³⁰⁾

In addition, the antibacterial mechanism of AgNPs is also due to their ability to produce high levels of reactive oxygen species (ROS) and free radicals,⁽³¹⁻³³⁾ while suppressing the expression of antioxidant enzymes (glutathione [GSH], superoxide dismutase and catalase), thereby accelerating ROS accumulation.^(34,35) The released Ag⁺ from AgNPs can interact with respiratory chain proteins on the membrane, interrupt intracellular O₂ reduction, and induce ROS production.⁽³⁶⁾ In turn, the accumulation of ROS leads to an apoptosis-like response, lipid peroxidation, GSH depletion, and DNA damage.⁽³⁷⁻³⁹⁾

Silver nanoparticles' unique properties make them suitable catalysts in various chemical reactions. The enhanced antimicrobial and virucidal effects of AgNPs are explained by their increased reactivity. In the range of 1–100 nm, AgNPs are tens and hundreds of times more active than other known biocidal and antibiotic drugs and are considered more promising agents as an alternative to antibiotics.⁽⁴⁰⁾ In the available literature, we have not found data on the formation of resistance of microorganisms to AgNPs, which, according to several authors, opens the way to overcoming resistance to antibiotics.^(40,41)

At the same time, some studies have reported bacterial resistance to AgNPs. In a study by Kaweeteerawat et al.,⁽⁴²⁾ it was found that AgNPs enhanced bacterial resistance to antibiotics by promoting stress tolerance by inducing intracellular ROS. In addition, the gram-negative bacteria *Escherichia coli* 013 and *Pseudomonas aeruginosa* CCM 3955, and *E. coli* CCM 3954 can develop resistance to AgNPs after repeated exposure. This resistance was due to the production of flagellin, an adhesive protein of the bacterial flagellum, which caused the aggregation of AgNPs and thereby eliminated their antibacterial effect.⁽⁴³⁾ Genes encoding silver resistance were detected most frequently in *Enterobacter* spp. (48%), followed by *Klebsiella* spp. (41%) and *Escherichia coli* (4%).⁽⁴⁴⁾

It was initially believed that AgNPs exert their activity only by releasing ions, acting as a depot. Modern research indicates that both ions and nanoparticles themselves exhibit activity.^(45,46) Their pharmacological activity is based on the ability of silver to be an electron pair acceptor.^(47,48) It has been demonstrated that the cellular membrane of bacteria has a negative charge due to the presence of carboxyl, phosphate, and amino groups.⁽⁴⁹⁾ The bactericidal efficiency of various positively and negatively charged silver nanoparticles has been extensively evaluated in the literature. The positively

charged silver nanoparticles showed the highest bactericidal activity against all microorganisms tested.⁽⁵⁰⁾

Thus, the interaction of Ag⁺ with a bacterial cell is complex and multifactorial.⁽⁵¹⁾ Since the mechanism of action of AgNPs is not specific, AgNPs act almost equally on both gram-positive and gram-negative microflora.⁽¹⁾ Due to the multi-component mechanism of action, microorganisms' resistance to Ag preparations rarely develops. The antifungal activity of silver preparations has also been described, with a predominantly positive mechanism of action in the form of cell membrane destruction.

In addition to antimicrobial and antifungal effects, Ag-based drugs are characterized by local anti-inflammatory activity.⁽⁵²⁾ In contrast to the antimicrobial effect, the anti-inflammatory effect of silver at the molecular level has not been sufficiently studied.⁽⁵¹⁾ The acceleration of the transition of a wound from inflammation to regeneration in experiments on animal models is associated with an increase in the concentration of epithelial growth factor after the application of silver dressing to the wound.⁽⁸⁾ Studies on human cells have found that application of silver at concentrations of 10–20 µg/ml causes a decrease in TNF-α, IL, and IL-6.⁽⁵³⁾

Despite significant clinical experience with the use of Ag-based compounds in medical practice, the results of scientific studies of their effectiveness and safety have been regularly subjected to critical evaluation.⁽⁵⁴⁾ One potential adverse reaction of using silver preparations with local exposure to wounds is the risk of developing argyria.⁽⁵⁵⁾ It should also be noted that there may be a deposition of silver in the cutaneous scar tissue and a change in skin color with prolonged use of the silver dressing.⁽⁵⁶⁾ In vitro studies have shown that there is a difference between the number of fibroblasts and the amount of collagen they produce when exposed to silver and non-silver dressings. Silver dressings reduce the number of fibroblasts by 54%–70% and collagen by 48%–68%.⁽⁵⁷⁾

The study of the effectiveness of Ag⁺ in antimicrobial agents has resumed. Published results of randomized clinical trials about the topical use of Ag-based preparations in the treatment of pressure ulcers,⁽⁵⁸⁾ microbially contaminated wounds with a high risk of infection,^(59,60) acute wounds during vacuum therapy,⁽⁶¹⁾ and colorectal⁽⁶²⁻⁶⁴⁾ and plastic surgery^(65,66) indicate their safety and clinical effectiveness.

A wide range of studies have focused on the use of AgNPs for wound healing. The use of AgNPs in ointments and dressings⁽⁶⁷⁻⁷⁰⁾ for the purpose of preventing wound infection⁽⁷¹⁾ and treating infected wounds⁽⁷²⁾ is said to be mandatory in many works. Many authors note the effectiveness of AgNPs as antibacterial agents,^(73,74) as well as nanocomposites, which include silver in nanosized form^(75,76) and nanofibers.^(77,78) Some researchers have been able to combine the antibacterial and proliferative effects of Ag-nanocomposites. In burn wounds, they have been shown to induce the proliferation and migration of keratinocytes and fibroblasts, promoting accelerated wound healing in diabetic mice.^(79,80) It was found that AgNPs stabilized with chitosan and polyvinyl alcohol not only inhibit the growth of bacteria but also can accelerate wound repair.^(81,82)

In addition, AgNP targets mesenchymal stem cell proliferation and osteogenic differentiation in vitro.⁽⁸³⁾ Mouse

models of burn wounds have demonstrated pronounced anti-inflammatory properties of polyamidoamine dendrimer-stabilized AgNPs.⁽⁸⁴⁾

One of the most important problems is the synthesis of sufficiently stable AgNPs of a given size that retain high chemical or biological activity for a long time. In this regard, when developing methods for synthesizing AgNPs, much attention is paid to the choice of stabilizers, which determine the stability level of AgNPs.⁽⁸⁵⁾ One of the possible solutions to this problem was using sodium alginate, a biopolymer of plant origin.⁽⁸⁶⁾ It has been shown that sodium alginate macromolecules reduce Ag⁺ into small nanoparticles and simultaneously stabilize them. It has been observed that one composition not only combines the properties of two biologically active substances but also their synergy.⁽⁸⁶⁾

The data analysis demonstrates the effectiveness of AgNPs as antibacterial and anti-inflammatory agents. Silver compounds have been shown to inhibit the growth and formation of bacterial biofilms.⁽⁸⁷⁾ In addition, it has been shown that the use of silver compounds in the treatment of wounds is also characterized by an improvement in the quality of life of patients and good cost-effectiveness. From this perspective, the development and introduction into clinical practice of new AgNP-based drugs for treating wounds and trophic ulcers, and the search for enhancing their bactericidal effect, is extremely relevant.

Competing Interests

The authors declare that they have no competing interests.

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